

ABSTRACT:

Transient-state methods to determine all the mass/charge transport properties of an inorganic compound

H.-I. Yoo

Seoul National University, Seoul 08826, Korea

All the mass/charge transport properties of an inorganic compound with single-type ions(i) and holes(h) as mobile charged components, e.g., $\{O^{2-},h^+\}$, may be exhaustively and succinctly documented in terms of a coupling coefficient matrix L of the Onsagarian causality as [1]

(\mathbf{I}) (\mathbf{I})	т	T)	$(-\nabla \eta_i)$	
$\begin{pmatrix} \mathbf{J}_i \\ \end{pmatrix} = \begin{pmatrix} \mathbf{L}_{ii} \end{pmatrix}$	L_{ih}	L _{iT}	$-\nabla n$	
$ \begin{pmatrix} \mathbf{J}_i \\ \mathbf{J}_h \end{pmatrix} = \begin{pmatrix} \mathbf{L}_{ii} \\ \mathbf{L}_{hi} \end{pmatrix} $	L_{hh}	L_{hT}	• • Ih	•
	1111	111 /	$\left(-\nabla T \right)$	

where, J_k and η_k denote the flux and electrochemical potential of the mobile charged-component k(=i,h), respectively, and T the absolute temperature. Due to the Onsager reciprocity [2] and the L-matrix transformation rule [1], respectively,

<-- \

$$\mathbf{L}_{ih} = \mathbf{L}_{hi} \quad ; \quad \begin{pmatrix} \mathbf{L}_{iT} \\ \mathbf{L}_{hT} \end{pmatrix} = \begin{pmatrix} \mathbf{L}_{ii} & \mathbf{L}_{ih} \\ \mathbf{L}_{hi} & \mathbf{L}_{hh} \end{pmatrix} \begin{pmatrix} \overline{\mathbf{S}}_i \\ \overline{\overline{\mathbf{S}}}_h \end{pmatrix}$$

where, due to Wagner, [3] $\overline{\overline{S}}_k$ is the transported entropy of k, the sum of its partial entropy, \overline{S}_k and entropy-of- transport, S_k^* or

$$\overline{\overline{S}}_{k} \equiv \overline{S}_{k} + S_{k}^{*} ; S_{k}^{*} \equiv q_{k}^{*} / T$$

with q_k^* being the reduced heat-of-transport of k(=i,h).

In this talk, we will introduce the transient-state methods to determine once and for all all the mass/charge transport properties $\{L_{ii}, L_{ih}(=L_{hi}), L_{hh}; L_{iT}, L_{hT}\}$, together with their experimental

DSL2023

implementations based on the isothermal semi-blocking polarizations and nonisothermal polarizations or thermopowers.

- [1] T. Lee, H.-S. Kim and H.-I. Yoo, Solid State Ionics, 262, 2 (2014).
- [2] D.-K. Lee and H.-I. Yoo, Phys. Rev. Lett., 97, 255901 (2006).
- [3] C. Wagner, Prog. Solid State Chem., 7, 1 (1972).